MATH 481, FALL 2024

PROJECT 2 DETECTING ART FORGERIES

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ABSTRACT. We formulate a system of differential equations corresponding to evolution in time of a mixture of radioisotopes uranium-238, radium-226, lead-210, and their eventual conversion to lead-206 which is not radioactive. We explain how our analysis may be applied to detect a forged oil painting.

1. INTRODUCTION

Say what this article is about. You may want to refer to the presentation in Martin Braun's book [1], the *Wikipedia* articles [2, 3], and a plethora of articles on Johannes Vermeer and Han van Meegeren on the web.

Please note that a recounting of the subject's history may be useful to establish a context, but that is certainly inadequate as an introduction to the subject of the current project. An article's introduction should tell the reader what is to be found *in this article*, as in "In this article we explain a method of distinguishing between a genuine and forged oil paintings through the analysis of the ratios of radium-226 and lead-210 isotopes which are generally present in all lead-based oil paints".

A further detailed overview can help, such as: "In Section 2 we derive a system of differential equations that describe the quantities of radium-226 and lead-210 as functions of time." Then "In Section ... we do ...".

2. The mathematical model

A radioactive material, such as uranium-238 (chemical symbol 238 U), decays according to:

$$\frac{d}{dt}Q(t) = -\lambda Q(t),$$

where Q(t) is the amount of the radioactive material at time t and λ is the material's *decay rate* or *decay constant*. In the context of this project, the "amount of material" refers to the number of atoms present in the sample.

Ideas for your writeup:

- Explain what a half-life is and derive the relationship between the decay rate λ and half-life τ .
- Why are ²³⁸U, ²²⁶Ra and ²¹⁰Pb always found together?
- The half-lives of ²³⁸U, ²²⁶Ra and ²¹⁰Pb are 4.5 billion years, 1600 years and 22 years, respectively. Compute their decay rates, λ_u , λ_r , and λ_p .

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- Let \overline{U} be the amount of ²³⁸U present in a sample. The decay rate of ²³⁸U is so small that over the span of a few thousand years its amount effectively is a constant.
- Derive the differential equation for the amount of radium-226, R(t), in the sample.
- Derive the differential equation for the amount of lead-210, P(t), in the sample.
- Assume that radium-226 and lead-210 in the sample have reached mutual equilibrium with the uranium-238 that generates them. Show that their equilibrium amounts, \bar{R} and \bar{P} , are related to the amount of the uranium source, \bar{U} , according to

$$\lambda_u \bar{U} = \lambda_r \bar{R} = \lambda_p \bar{P},$$

or equivalently,

$$\frac{\bar{U}}{\tau_u} = \frac{\bar{R}}{\tau_r} = \frac{\bar{P}}{\tau_p},$$

where τ_u , τ_r , and τ_p are the half-lives of ²³⁸U, ²²⁶Ra, and ²¹⁰Pb, respectively.

3. Detecting a forgery

Consider lead ore that has been in an undisturbed state for many thousands of years. Then the amounts \bar{U} , \bar{R} , and \bar{P} of ²³⁸U, ²²⁶Ra and ²¹⁰Pb will be interrelated as in the equilibrium equations (1).

The ore is taken to the smeltery to extract its lead. Essentially all of lead-210 and lead-206 is recovered but most of uranium and radium goes out as waste. Thus the equilibrium among 238 U, 226 Ra and 210 Pb in the resulting product is lost.

Let's say that only a fraction α of the uranium and radium remains in the processed lead. For numerical experiments take $\alpha = 10^{-4}$. Write a new set of differential equations and initial conditions that models the decay of the isotopes in the new product. Solve the system of differential equation for R(t) and P(t). Find the expression for the ratio $\rho(t) = P(t)/R(t)$. Verify that the ratio converges to $\overline{P}/\overline{R}$ as $t \to \infty$. Plot $\rho(t)$ over a span of 300 years. Explain the relevance of this to detecting an art forgery.

You may have use for Figure 1 as you develop your article's mathematical model. Note that there is no external graphes. The entire graphics is defined within the LAT_EX file.

References

- Martin Braun. Differential equations and their applications. 2nd ed. An introduction to applied mathematics; Applied Mathematical Sciences, Vol. 15. Springer-Verlag, 1978.
- [2] Wikipedia. Johannes Vermeer. Sept. 2024. URL: http://en.wikipedia.org/ wiki/Johannes_Vermeer.
- [3] Wikipedia. Uranium-238. Sept. 2024. URL: http://en.wikipedia.org/wiki/ Uranium-238.

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REFERENCES

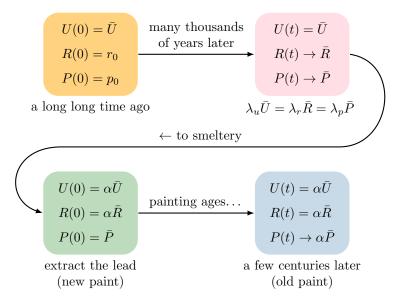


FIGURE 1. Your caption here.