#### **University of Maryland Baltimore County - UMBC**

Phys650 - Special Topics in Experimental Atmospheric Physics (Spring 2009)

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# Class 5

- In Depth Discussion and Continuation of the Course Project
- Introduction to Radiative Transfer: Theory and Codes
- Introduction to radiometry
- Hands on Experiment: Calibration and Characterization of the Sunphotometer

#### In depth discussion and continuation of the course project:

- Introduction to Radiative Transfer and Radiative Transfer codes
- Introduction to Radiometric Analysis:
  - How many photos arrive at the surface of your detector?
  - What is the uncertainty of your measurement?
- Discussion on the FOV of the Photometer
  - What is the FOV of your system?
  - What is the FOV of the sun on the sky?
  - What is the associated error from the difference between the above FOVs?
- Discussion on the Calibration of the Photometer
  - What is the required accuracy for sunphotometry and what method can meet this requirement?
  - Calibration with Langley Plots
  - Calibration in Integrating Spheres

After calibration, you can start measuring the aerosol optical depth at some location and compare it with Aeronet measurements, Lidar, or satellite retrievals. Notice that you cannot touch/modify your photometer after calibration.

- Discussion on measurements of sky radiance, moon photometry, etc.
  - How can you convert your sunphotometer to a sky radiometer?
  - How many photos do you get from the sky at the surface of your detector?
  - Can you convert your sunphotometer to a moon, or a star photometer?
  - How many photons do you need arriving at the surface of your detector in order to make an accurate measurement in each case?
  - How can you increase the number of photons arriving at the surface of your detector

#### Experiments

#### 12:00-12:30 - Hands on Experiment:

- Assemble a "Tiger team" to design your instrument and attack the questions above. Each team should address all the questions above theoretically and experimentally. Try to explore the expertise of the team. You should understand the whole experiment but, if you are better in theory or in computation, go for it and help the team to achieve its goals!!!
- Visit to the Integration sphere facility. Experiments with the sphere should be planned in advance. The sphere can only be used under supervision.

#### 12:30- 3:00 PM - Open Lab:

Tiger Teams shall design and start their experimental setups to measure the properties of their photometers. Measurements will include:

- Measurements of the LED FOV (emission and detection)
- Alignment of the LED detection via emission
  - I have noticed a dark spot at the center of some LEDS!!! Double check if it have a negative effect on the detection!!!
- Radiometric calibration at the integrating sphere

#### Radiance sources and sinks affecting visible and IR wavebands



### Solar Energy Paths





diffuse + direct

direct + diffuse

multiple scattering

#### Solar (reflective) spectral domain



## **Observation Geometry**







### Perfect Lambertian Reflector

Radiance of the Perfect Lambertian Reflector





 $\rho_{Perfect \ Lambertian \ reflector}(\theta_S, \theta_V, \phi) = 1$ 

$$\rho_{Lambertian reflector}(\theta_{S}, \theta_{V}, \phi) = \rho$$

## **Different Types of Reflectors**



Specular reflector (mirror)



Nearly Specular reflector (water)



diffuse reflector (lambertian)



nearly diffuse reflector



Hot spot reflection



#### **SRTE 2 interactions**



#### **SRTE Multiple Interactions**

$$\rho_{app} = \rho_{atm} + T(\theta_{s})T(\theta_{v})\rho_{ground} \left[ l + \rho_{ground}S_{atm} + (\rho_{ground}S_{atm})^{2} + (\rho_{ground}S_{atm})^{3} \dots \right]$$

$$l + r + r^{2} + r^{3} + \dots r^{n-1} = \frac{l - r^{n}}{l - r}$$

 $\rho_{ground}S_{atm}$  < 1 so when n->∞ then  $(\rho_{ground}S_{atm})^n$  ->0

**Therefore** 
$$\left[1 + \rho_{ground}S + (\rho_{ground}S)^2 + (\rho_{ground}S)^3 \dots\right] = \frac{1}{1 - \rho_{ground}S}$$

$$\rho_{app} = \rho_{atm} + T_{atm}(\theta_s) T_{atm}(\theta_v) \frac{\rho_{ground}}{1 - \rho_{ground}} S_{atm}$$

#### BRDF atmosphere coupling correction

Lambertian infinite target approximation

$$\rho_{app}(\theta_s, \theta_v, \phi) = \rho_{atm}(\theta_s, \theta_v, \phi) + T_{atm}(\theta_s) T_{atm}(\theta_v) \frac{\rho_{ground}}{1 - \rho_{ground} S_{atm}}$$

**BDRF** atmosphere coupling approximation

$$\begin{split} \rho_{app}(\theta_{s},\theta_{v},\phi) &= \rho_{atm}(\theta_{s},\theta_{v},\phi) + e^{-\tau/\mu_{s}}e^{-\tau/\mu_{v}}\rho_{s}(\theta_{s},\theta_{v},\phi) \\ &+ e^{-\tau/\mu_{v}}t_{d}(\theta_{s})\overline{\rho}_{s} + e^{-\tau/\mu_{s}}t_{d}(\theta_{v})\overline{\rho}'_{s} + t_{d}(\theta_{v})t_{d}(\theta_{s})\overline{\overline{\rho}}_{s} \\ &+ \frac{T_{atm}(\theta_{s})T_{R+A}(\theta_{v})S_{atm}(\overline{\overline{\rho}}_{s})^{2}}{1-S_{atm}\overline{\overline{\rho}}_{s}} \\ &\quad \overline{\rho}_{s}(\mu_{s},\mu_{v},\phi) = \frac{\int_{0}^{2\pi}\int_{0}^{1}\mu L_{atm}^{\downarrow}(\mu_{s},\mu,\phi')\rho_{s}(\mu,\mu_{v},\phi'-\phi)d\mu d\phi'}{\int_{0}^{2\pi}\int_{0}^{1}\mu L_{atm}^{\downarrow}(\mu_{s},\mu,\phi')d\mu d\phi'} \\ &\quad \overline{\overline{\rho}}_{s}(\mu_{s},\mu_{v},\phi) = \overline{\overline{\rho}'_{s}(\mu_{s},\mu_{v},\phi)} \qquad \overline{\rho}'_{s}(\mu_{s},\mu_{v},\phi) = \overline{\rho}_{s}(\mu_{s},\mu_{v},\phi) \end{split}$$

# STRE for non absorbing atmosphere and lambertian ground



# The composition of the atmosphere

Permanent Constituents

Variable constituents

Constituent	% by	Constituent	% by volume
	volume		5
Nitrogen $(N_2)$	78.084	Water Vapor $(H_2O)$	0.04
Oxygen $(O_2)$	20.948	Ozone $(O_3)$	12 x 10 <sup>-4</sup>
Argon (Ar)	0.934	Sulfur dioxide $(SO_2)^b$	0.001 x 10 <sup>-4</sup>
Carbon dioxide $(CO_2)$	0.033	Nitrogen dioxide ( $\tilde{NO}_2$ )	0.001 x 10 <sup>-4</sup>
Neon (Ne)	18.18 x 10 <sup>-4</sup>	Ammonia (NH <sub>3</sub> )	0.001 x 10 <sup>-4</sup>
Helium (He)	5.24 x 10 <sup>-4</sup>	Nitric oxide (NO)	0.0005 x 10 <sup>-4</sup>
Krypton (Kr)	1.14 x 10 <sup>-4</sup>	Hydrogen sulfide $(H_2S)$	0.00005 x 10 <sup>-4</sup>
Xenon (Xe)	0.089 x 10 <sup>-4</sup>	Nitric acid vapor	trace
Hydrogen $(H_2)$	$0.5 \times 10^{-4}$	•	
Methane $(CH_4)$	1.5 x 10 <sup>-4</sup>		
Nitrous Oxide $(N_2O)$	0.27 x 10 <sup>-4</sup>		
Carbon Monoxide (CO)	0.19 x 10 <sup>-4</sup>		

## Gaseous Absorption (H<sub>2</sub>O)



by E. Vermote et al., University of Maryland/ Dept of Geography

# Modified SRTE to account for absorption

In case of a pure molecular atmosphere (no aerosol) we can write:

$$\rho_{app}(\theta_s,\theta_v,\phi) = Tg^{other\,gases}(m,U_{gaz}) \left[ \rho_{atm}(\theta_s,\theta_v,\phi) + Tg^{H_2O}(m,U_{H_2O})T_{atm}(\theta_s)T_{atm}(\theta_v) \frac{\rho_{ground}}{1 - S_{atm}\rho_{ground}} \right]$$

m is the air mass =  $1/\cos(\theta_s) + 1/\cos(\theta_v)$ U<sub>gaz</sub> is the gaz concentration

## Phase function

• The phase function,  $P(\Theta)$ , describe the distribution of scattered radiation for one or an set of particles. It is normalized such as:  $\int P(\Theta) d\omega = 4\pi$ 

since

$$\int_{0}^{2\pi\pi} \int_{0}^{2\pi\pi} P(\Theta) \sin(\theta) d\theta d\phi = 2\pi \int_{0}^{\pi} P(\theta) \sin(\theta) d\theta$$

 $\int_{0}^{\pi} P(\theta) \sin(\theta) d\theta = 2$ 

# Rayleigh/molecular scattering 1/4

• Rayleigh or molecular scattering refers to scattering by atmospheric gases, in



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# Aerosol scattering 2/5





Scattering Angle (deg)

#### Some Useful Radiative Transfer codes:

#### Mie codes:

Homogeneous Sphere: <u>ftp://climate1.gsfc.nasa.gov/wiscombe/Single\_Scatt/Homogen\_Sphere/Exact\_Mie</u> Layered Sphere: <u>ftp://climate1.gsfc.nasa.gov/wiscombe/Single\_Scatt/Coated\_Sphere</u>

#### **Disort:**

For Fortran 77: <u>ftp://climate1.gsfc.nasa.gov/wiscombe/Multiple\_Scatt/DISORT1.2</u>

For Fortran 90: <u>ftp://climate1.gsfc.nasa.gov/wiscombe/Multiple\_Scatt/DISORT1.3</u>

Other Multiple Scattering Codes: <u>ftp://climate1.gsfc.nasa.gov/wiscombe/Multiple\_Scatt</u>

SBDART: <a href="http://www.paulschou.com/tools/sbdart/">http://www.paulschou.com/tools/sbdart/</a>