Homework 1

Ideal gas worksheet

Name:

The figure below shows the P-V diagram for **two** moles of an ideal monatomic gas that expands and contracts along an elliptical path.







Q1: Which shaded region (A, B, C, D) represents the work done by the gas as it expands from 3L to 7L?

Q2: Which shaded region (A, B, C, D) represents the work done by the gas as it contracts from 7L to 3L?

Q3: Which shaded region (A, B, C, D) represents the work done by the gas as it completes one cycle?

Q4: How many joules are equivalent to 100 kPa-L?

Q5: How many joules are equivalent to 1 kPa-L?

Q6: What is the general relationship between kPa-L and joules?

Q7: What is the temperature of the gas at the uppermost point on the ellipse?



Q8: At which of the points (a,b,c,d) in the figure below is the gas temperature the highest?

Q9: The general equation for an ellipse in P-V space has the form:

$$\left(\frac{P-\alpha}{\beta}\right)^2 + \left(\frac{V-\chi}{\delta}\right)^2 = 1^{-1}$$

What are the values of α, β, χ and δ that describe the ellipse illustrated in the figures?

Q10: Using the results of Q9, write down an equation for P vs. V that describes the path illustrated *while the gas expands*.

Q11: Using the results of Q9, write down an equation for P vs. V that describes the path illustrated *while the gas contracts*.

Q12: Compute the work done by the gas as it expands from 3L to 7L.

Q13: Compute the work done by the gas as it contracts from 7L to 3L.

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Q14: Compute the work done by the gas as it completes one full cycle.

O11. Using the results of Q6, write down an equation for Pive. V that describes the path liustrated while the gas contracts.

312: Compute the work done by the gas as it expands from 31 to 71